### **Preliminary communication**

## SYNTHESIS AND CHARACTERIZATION OF A DIMERIC TRIS(ARSINO)GALLANE CONTAINING A NONPLANAR (Ga-As)<sub>2</sub> RING: CRYSTAL STRUCTURE OF {[(Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>As]<sub>3</sub>Ga}<sub>2</sub> \*

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#### Summary

The dimer  $\{[(Me_3SiCH_2)_2As]_3Ga\}_2$ , only the second tris(arsino)gallane to be completely characterized, has been prepared by the reaction of  $(Me_3SiCH_2)_2AsLi$  with GaCl<sub>3</sub>; X-ray crystallographic analysis shows it to be the first example of a compound containing a distinctly nonplanar four-membered ring of alternating four-coordinate Ga and As atoms.

Recently, we applied two new synthetic methods to the preparation of compounds containing a gallium-arsenic bond, viz., dehalosilylation between a silylarsine and a halogallane [1], and coupling using a lithium arsenide and a chlorogallane [2]. Among the compounds prepared by both methods is the first example of a tris(arsino)gallane, monomeric (Mes<sub>2</sub>As)<sub>3</sub>Ga, which X-ray analysis [2] has shown to contain three-coordinate gallium and arsenic. Subsequently, (Bu<sup>1</sup><sub>2</sub>As)<sub>2</sub>Ga was reported by others, but data for a crystal structure were not obtainable [3]. We now report the structure of a second tris(arsino)gallane, dimeric [(Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>As]<sub>3</sub>Ga (1), prepared by the lithium arsenide method. Interestingly, as noted previously, the reaction of (Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>AsSiMe<sub>3</sub> with GaCl<sub>3</sub> did not yield 1 [1]. Compound 1 has a solid state structure containing a distinctly nonplanar four-membered ring of alternating four-coordinate Ga and As atoms. This form contrasts with the planar, centrosymmetric (Ga-As)<sub>2</sub> units in [(Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>AsGaPh<sub>2</sub>]<sub>2</sub> (2) [4], the first dimeric mono(arsino)gallane for which the structure was reported, and in  $(Bu_2^tAsGaMe_2)_2$  (3) [3], and the nearly planar unit in  $(Bu_2^tAsGaBu_2)_2$  (4) [3], but is similar to, although less puckered than, the novel nonplanar  $(Ga-S)_2$  form found

<sup>\*</sup> Dedicated to Professor G.E. Coates on the occasion of his 70th birthday.

in  $(Pr^{i}SGaI_{2})_{2}$  which contains two four-coordinate Ga atoms and two three-coordinate S atoms [5].



A suspension of  $(Me_3SiCH_2)_2AsLi$  [6] (2.03 g, 7.9 mmol) in hexane was added [7] to a hexane solution of GaCl<sub>3</sub> (0.46 g, 2.6 mmol) at  $-78^{\circ}C$ . After 18 h at room



Fig. 1. Molecular structure of {[( $Me_3SiCH_2$ )\_2As]\_3Ga}\_2 (1). Selected distances (Å) and angles (°) are: Ga-As(1) 2.581(1), Ga-As(2) 2.478(2), Ga-As(3) 2.476(2), Ga-As(1') 2.540(1), Ga'-As(1') 2.540(1), Ga'-As(2') 2.470(1), Ga'-As(3') 2.474(2), Ga'-As(1) 2.559(1), As(1)-Ga-As(1') 83.58(4), As(1)-Ga'-As(1') 84.04, Ga-As(1)-Ga' 94.57(4), Ga-As(1')-Ga' 96.02(4), As(2)-Ga-As(3) 122.37(5), As(2')-Ga'-As(3') 113.68(5), C(111)-As(1)-C(121) 103.0(4), C(111')-As(1')-C(121') 104.7(4).



Fig. 2. The nonplanar (Ga-As)<sub>2</sub> ring of compound 1.

temperature, the brown mixture formed was filtered, and the filtrate was evaporated and the residue dissolved in hexane. Crystallization  $(-78^{\circ}C)$  and cold filtration, followed by solvent removal, recrystallization, hexane washings, and finally drying in vacuo afforded {[(Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>As]<sub>3</sub>Ga}<sub>2</sub> (1) as a pale yellow solid (0.46 g, 22% yield) m.p. 71-149°C (dec.) [8]. Crystals suitable for an X-ray structure determination were grown from a C<sub>6</sub>F<sub>6</sub> solution [9].

Crystals of 1 comprise discrete centrosymmetrically-related dimers having the structure illustrated in Fig. 1. Several features of this dimer attest to its highly strained nature. Thus, the Ga-As(1)-Ga'-As(1') ring, with a dihedral angle of 13.6° (vs. 36.7(2)° in the (Ga-S)<sub>2</sub> ring of (Pr<sup>i</sup>SGaI<sub>2</sub>)<sub>2</sub>) between the As(1)-Ga-As(1') and As(1)-Ga'-As(1') planes (mean endocyclic dihedral angle about the ring bonds 10.2°) is, as shown in Fig. 2, distinctly non-planar. Two of the ring bonds, Ga-As(1') and Ga'-As(1') at 2.540(1) Å, are equal, and significantly shorter than the other pair, 2.559(1) and 2.581(1) Å, of which the latter is the longest distance yet reported for such a bond and contrasts with the corresponding longest values of 2.530(1), 2.558(1), 2.557(3), and 2.553(1) Å, respectively, for four-coordinate Ga in dimers 2, 3, and 4, and the unusual  $[(PhAsH)(R_2Ga)(PhAs)_6(RGa)_4]$  (R =  $Me_3SiCH_2$ ) cluster [10]. All the ring bonds of 1 are longer than the mean of the essentially equal exocyclic Ga-As bonded distances to three-coordinate As atoms, which, at 2.475 Å, is slightly shorter than the mean Ga-As distance for trigonal planar Ga in monomeric (Mes<sub>2</sub>As)<sub>1</sub>Ga. The mean ring bond angles in 1 (84.81° at Ga, 95.30° at As) are similar to those in dimers 2, 3, and 4 (range: 84.31-85.08° at Ga; 94.92–95.69° at As), but the exocyclic As-Ga-As angles involving the threecoordinate As atoms (122.37(5), 113.68(5)°) differ significantly in response to the different intramolecular interactions involving substituents at each of the Ga centers. Corresponding exocyclic C-As-C angles show much less variation (103.0(4), 104.7(4)°), indicating the greater resistance of the As centers to bond angle deformation.

Cryoscopic molecular weight determinations indicate that 1 remains intact as a dimer in solution at low temperatures. It appears, however, that the dimer is

fluxional in solution (the fluxional properties of a dimeric bis(arsino)gallane have been reported) [1], as indicated by broadening and eventual coalescence of <sup>13</sup>C NMR signals as the temperature is increased. Also, compound 1 is thermally unstable in solution at ambient temperature and above, and slowly decomposes to the diarsine  $[(Me_3SiCH_2)_2As]_2$  [1] and unknown products.

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- 6 (Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>AsLi was produced by the reaction of (Me<sub>3</sub>SiCH<sub>2</sub>)<sub>2</sub>AsH [4] and Bu<sup>n</sup>Li in hexane for 2 days at 60 °C, and isolated as an off-white powder.
- 7 All manipulations were performed under dry nitrogen.
- 8 Found: C, 35.34; H, 8.29; mol. wt.,  $1582 \pm 65$  (cryoscopic, 0.268 g in 12.22 g cyclohexane).  $C_{48}H_{132}As_6Ga_2Si_{12}$  calcd.: C, 35.25; H, 8.13%; mol. wt., 1636. <sup>1</sup>H NMR (300 MHz) ( $C_6D_6$ , 21°C):  $\delta$  0.32 (s, exo-Me<sub>3</sub>Si), 0.37 (s, endo-Me<sub>3</sub>Si), 1.32 and 1.79 (AB Pattern, <sup>2</sup>J(HH) 13.8 Hz, exo-CH<sub>2</sub>), 1.71 (endo-CH<sub>2</sub>); <sup>13</sup>C{<sup>1</sup>H} NMR (75.4 MHz) ( $C_6D_6$ , 21°C):  $\delta$  0.98 (s, exo-Me<sub>3</sub>Si), 2.02 (s, endo-Me<sub>1</sub>Si), 6.69 (s, exo-CH<sub>2</sub>), 10.59 (s, endo-CH<sub>2</sub>).
- 9 Crystal data:  $C_{48}H_{132}As_6Ga_2Si_{12}$  (1), M = 1635.59, triclinic, space group  $P\overline{1}$ , a 15.050(3), b 25.417(8), c 12.621(4) Å,  $\alpha$  93.73(3),  $\beta$  110.68(2),  $\gamma$  77.00(2)°, U 4400.5 Å<sup>3</sup>, Z = 2,  $D_c$  1.234 g cm<sup>-3</sup>,  $\mu$ (Cu- $K_{\alpha}$  radiation) 50.7 cm<sup>-1</sup>. The crystal structure was solved by direct methods. Full-matrix least-squares refinement of atomic positional and thermal parameters (anisotropic As, C, Ga, Si; fixed methylene H contributions) converged to R = 0.064 ( $R_w = 0.097$ ;  $w = 1/\sigma^2(|F_0|)$ ) over 8504 absorption-corrected reflections ( $I > 3.0 \sigma(I)$ ) recorded on an Enraf-Nonius CAD-4 diffractometer (Cu- $K_{\alpha}$  radiation,  $\lambda$  1.5418 Å; incident-beam graphite monochromator;  $\omega 2\theta$  scans,  $\theta_{max}$  57°). A table of atomic coordinates and a full list of bond lengths and angles has been deposited with the Cambridge Crystallographic Data Centre.
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